



GREEN  
CHEMISTRY

DSaT<sup>TM</sup>

Dipartimento Scienze Chimiche e Tecnologie dei Materiali

Young Investigator Award 2019

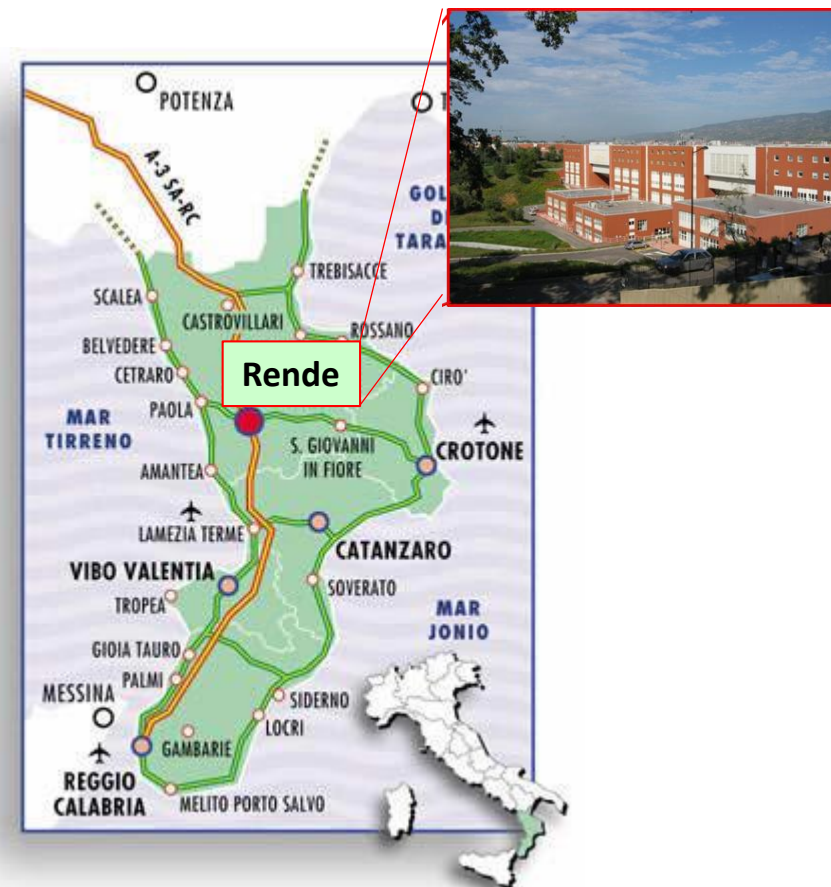


Consiglio Nazionale delle Ricerche

ITM

# Membranes for Green processes

Elisa Esposito



Consiglio Nazionale delle Ricerche

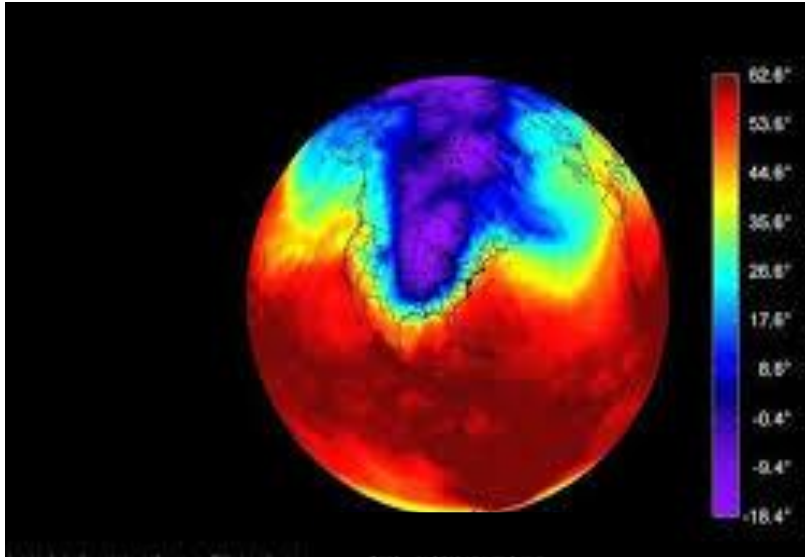


# Green Chemistry

**Green chemistry**, also called **sustainable chemistry**, is an area of chemistry and chemical engineering focused on the designing of products and processes that minimize or eliminate the use and generation of hazardous substances

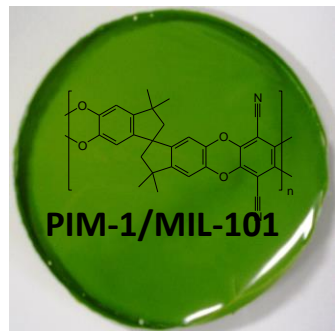
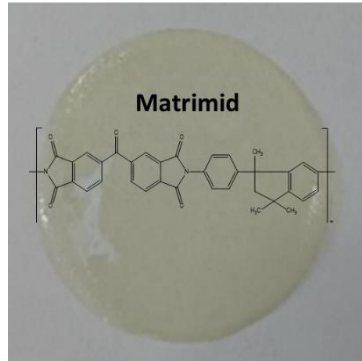


# Context



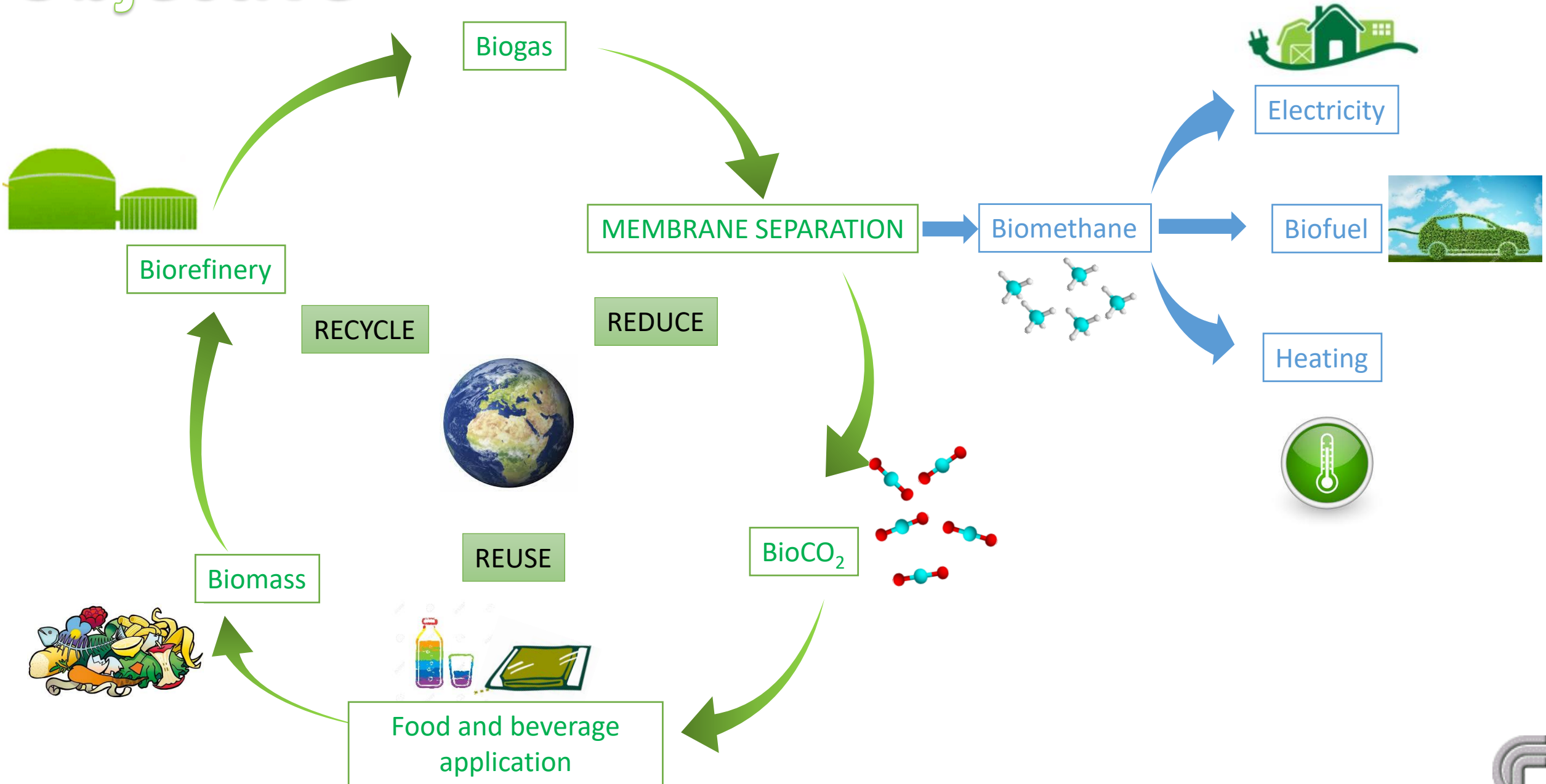
# Focus of Research

Design and development of new materials for membrane preparation





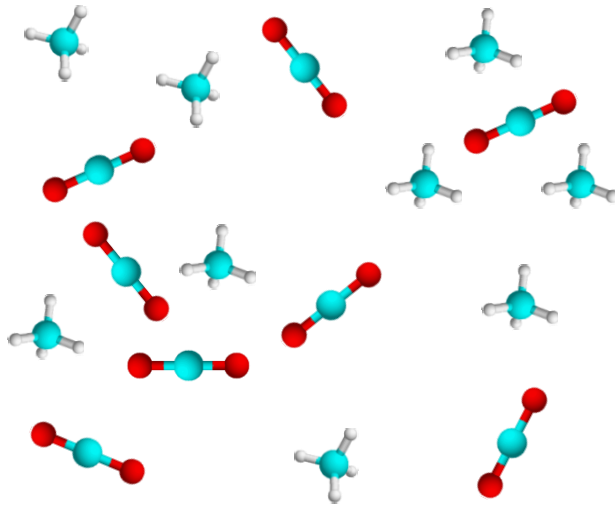
# Objective



# Gas transport in Membrane

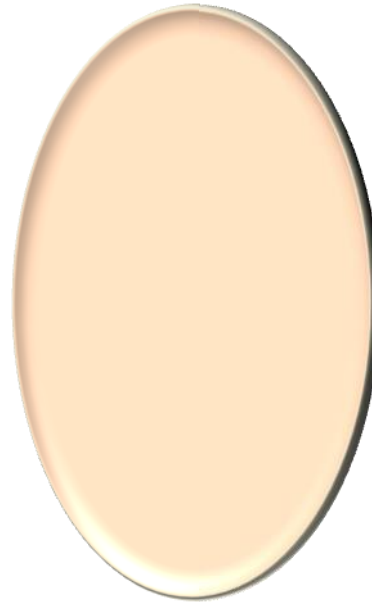
*Feed mixture*

e.g.  $\text{CO}_2/\text{CH}_4$



*Membrane*

(selective barrier)



*Retentate*

Enriched in methane

$\text{CH}_4 \gg \text{CO}_2$

$$\text{Permeability (P)} = \frac{n \cdot \ell}{t \cdot A \cdot \Delta P}$$

*Permeate*

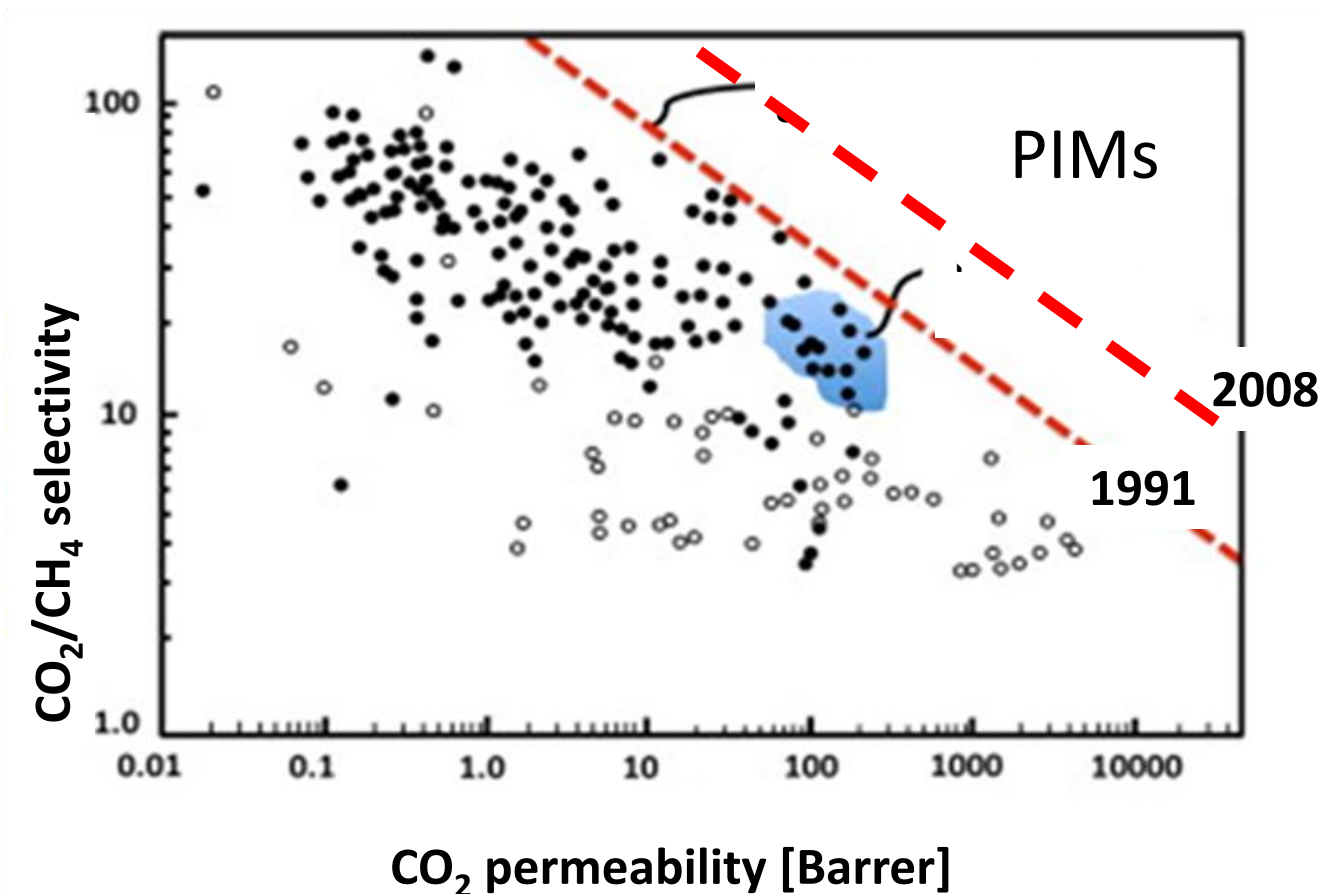
Enriched in  $\text{CO}_2$

$\text{CO}_2 \gg \text{CH}_4$

$$\text{selectivity } (\alpha_{AB}) = \frac{P_A}{P_B}$$



# Robeson plot



## Rubbery Polymers

  Permeability

 Selectivity

## Glassy polymers

 Permeability

  Selectivity

B.D. Freeman, Basis of permeability/selectivity tradeoff relations in polymeric gas separation membranes, *Macromolecules*. 32 (1999) 375–380  
L.M. Robeson, The upper bound revisited, *J. Memb. Sci.* 320 (2008) 390–400.



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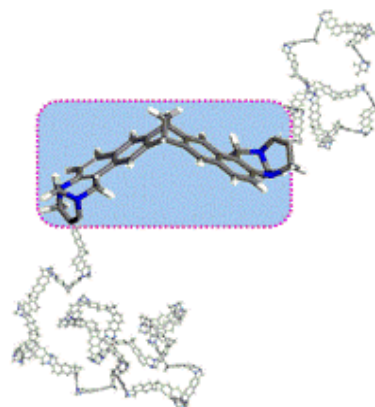
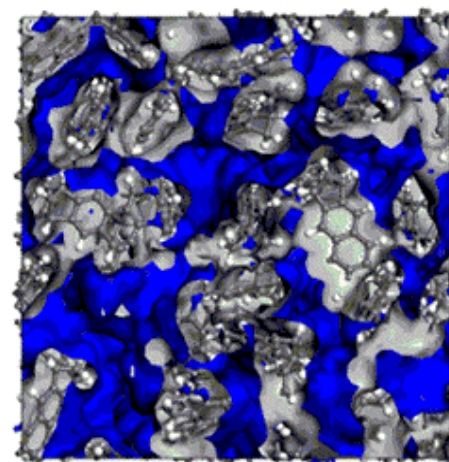
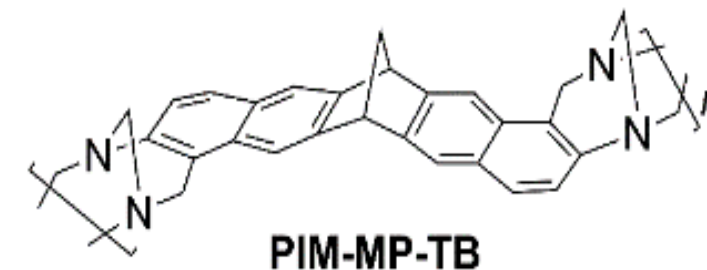


Cite this: *J. Mater. Chem. A*, 2018, 6, 5661

# A highly rigid and gas selective methanopentacene-based polymer of intrinsic microporosity derived from Tröger's base polymerization†

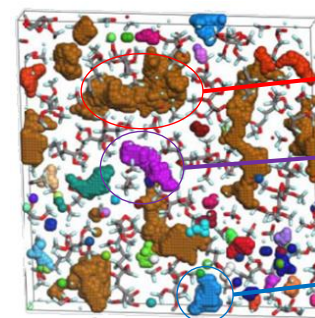
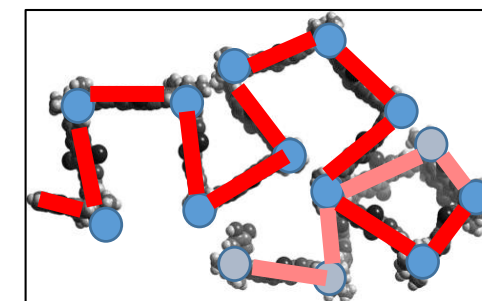
Rhodri Williams,<sup>a</sup> Luke A. Burt,<sup>a</sup> Elisa Esposito,<sup>b</sup> Johannes C. Jansen,<sup>id a,b</sup> Elena Tocci,<sup>id b</sup> Carmen Rizzuto,<sup>b</sup> Marek Lanč,<sup>id c</sup> Mariolino Carta,<sup>id \*d</sup> and Neil B. McKeown <sup>id \*\*</sup>

Polymers of intrinsic microporosity (PIMs) have been identified as potential next generation membrane materials for the separation of gas mixtures of industrial and environmental relevance. Based on the exceptionally rigid methanopentacene (MP) structural unit, a Polymer of Intrinsic Microporosity (PIM-MP-TB) was designed to demonstrate high selectivity for gas separations. PIM-MP-TB was prepared using a polymerisation reaction involving the formation of Tröger's base linking groups and demonstrated an apparent BET surface area of 743 m<sup>2</sup> g<sup>-1</sup> as a powder. The microporosity of PIM-MP-TB was also



Permeability

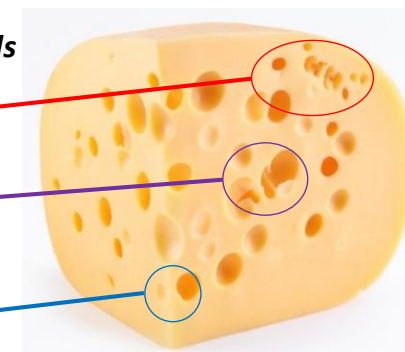
Selectivity



Highly interconnected voids

Interconnected Voids

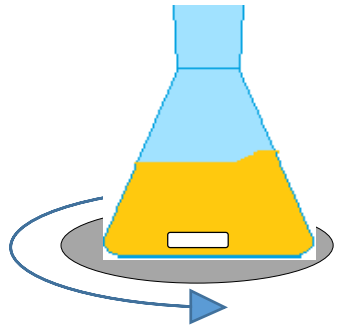
Single Voids



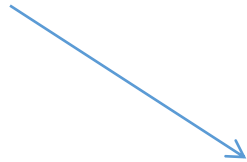


# Membrane preparation

PIM-MP-TB polymer solution



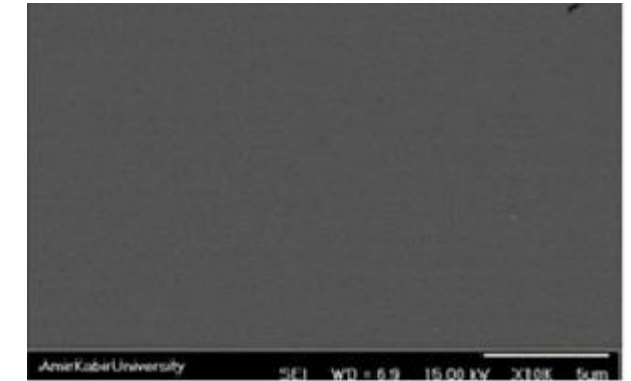
Mechanical stirring



Slow solvent evaporation

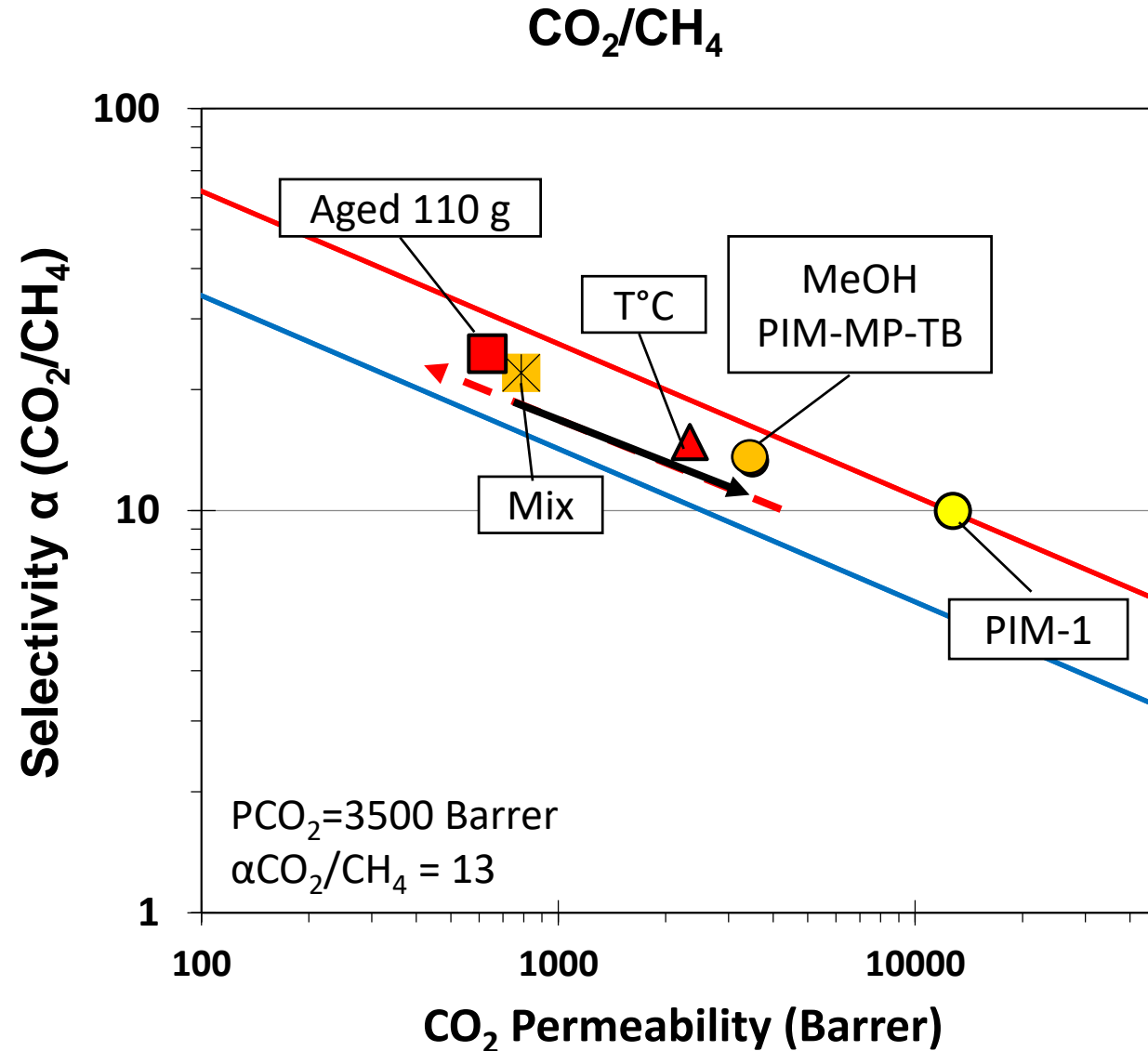


SEM image



Dense and homogenous membrane

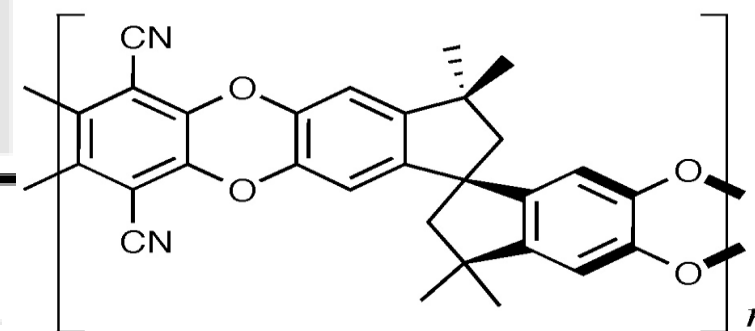
# Mixed gas permeation test





Contents lists available at ScienceDirect

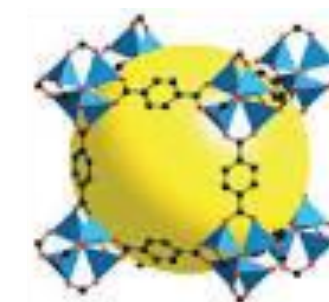
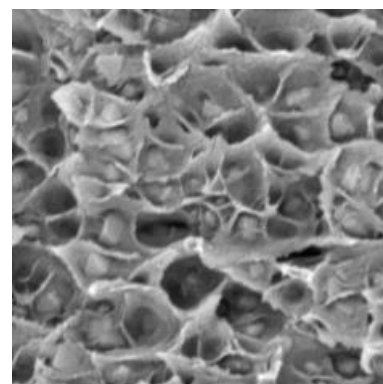
## Separation and Purification Technology

journal homepage: [www.elsevier.com/locate/seppur](http://www.elsevier.com/locate/seppur)

PIM-1

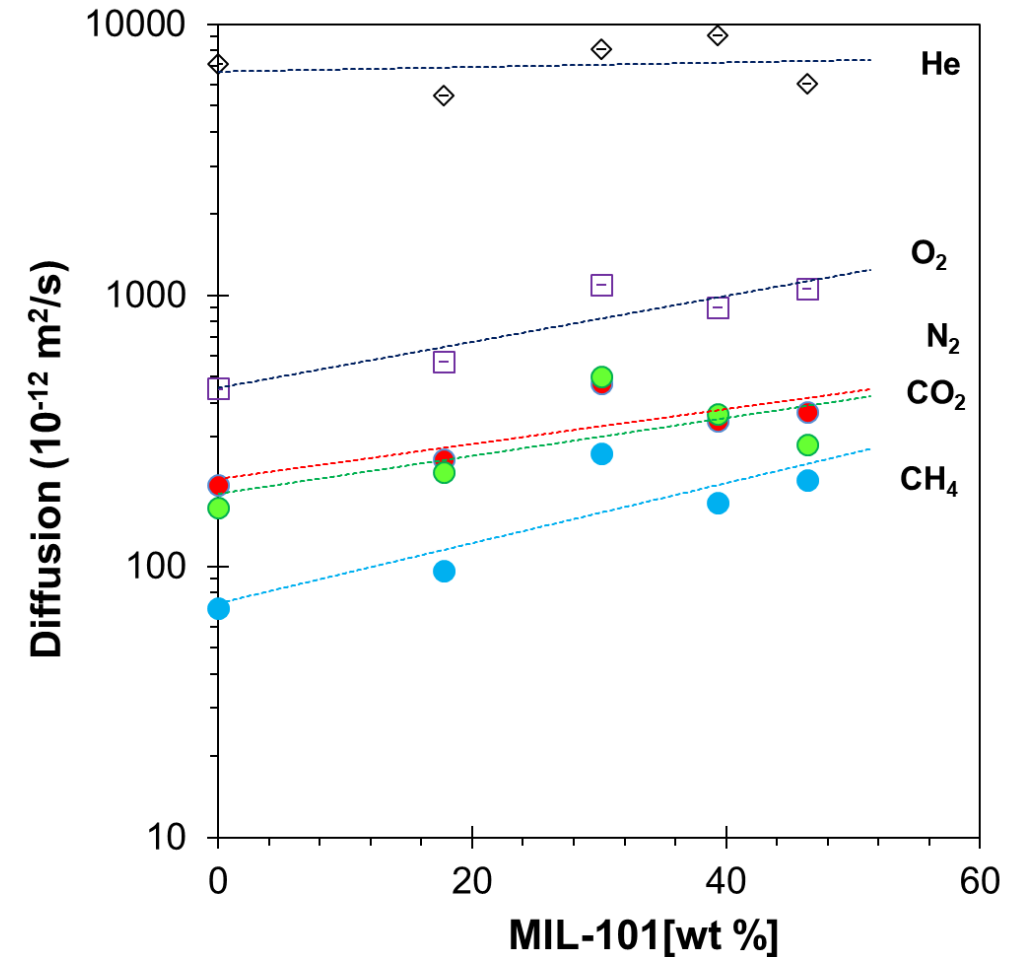
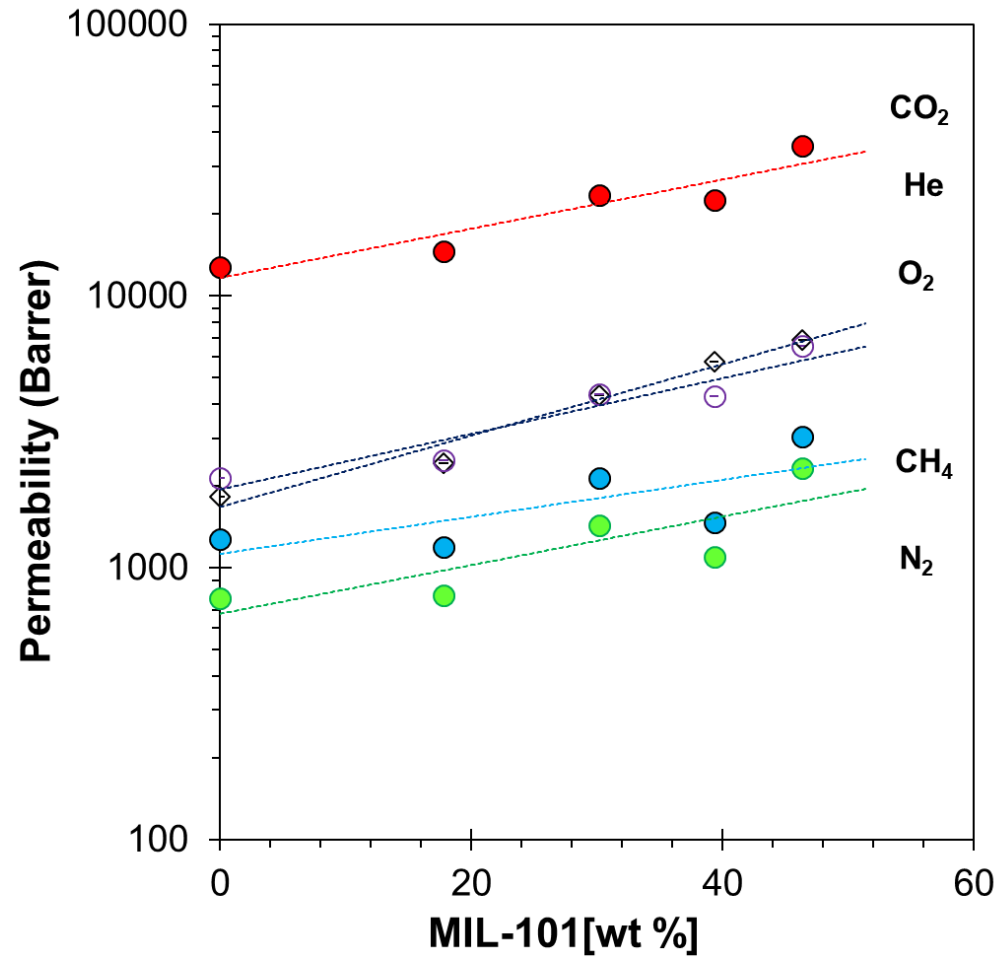
## Mixed matrix membranes based on MIL-101 metal–organic frameworks in polymer of intrinsic microporosity PIM-1

Muhammed Khedhayyer<sup>a</sup>, Alexandra F. Bushell<sup>a</sup>, Peter M. Budd<sup>a</sup>, Martin P. Attfield<sup>a,\*</sup>,  
 Dongmei Jiang<sup>b</sup>, Andrew D. Burrows<sup>b</sup>, Elisa Esposito<sup>c,\*</sup>, Paola Bernardo<sup>c</sup>, Marcello Monteleone<sup>c</sup>,  
 Alessio Fuoco<sup>c</sup>, Gabriele Clarizia<sup>c</sup>, Fabio Bazzarelli<sup>c</sup>, Amalia Gordano<sup>c</sup>, Johannes C. Jansen<sup>c</sup>

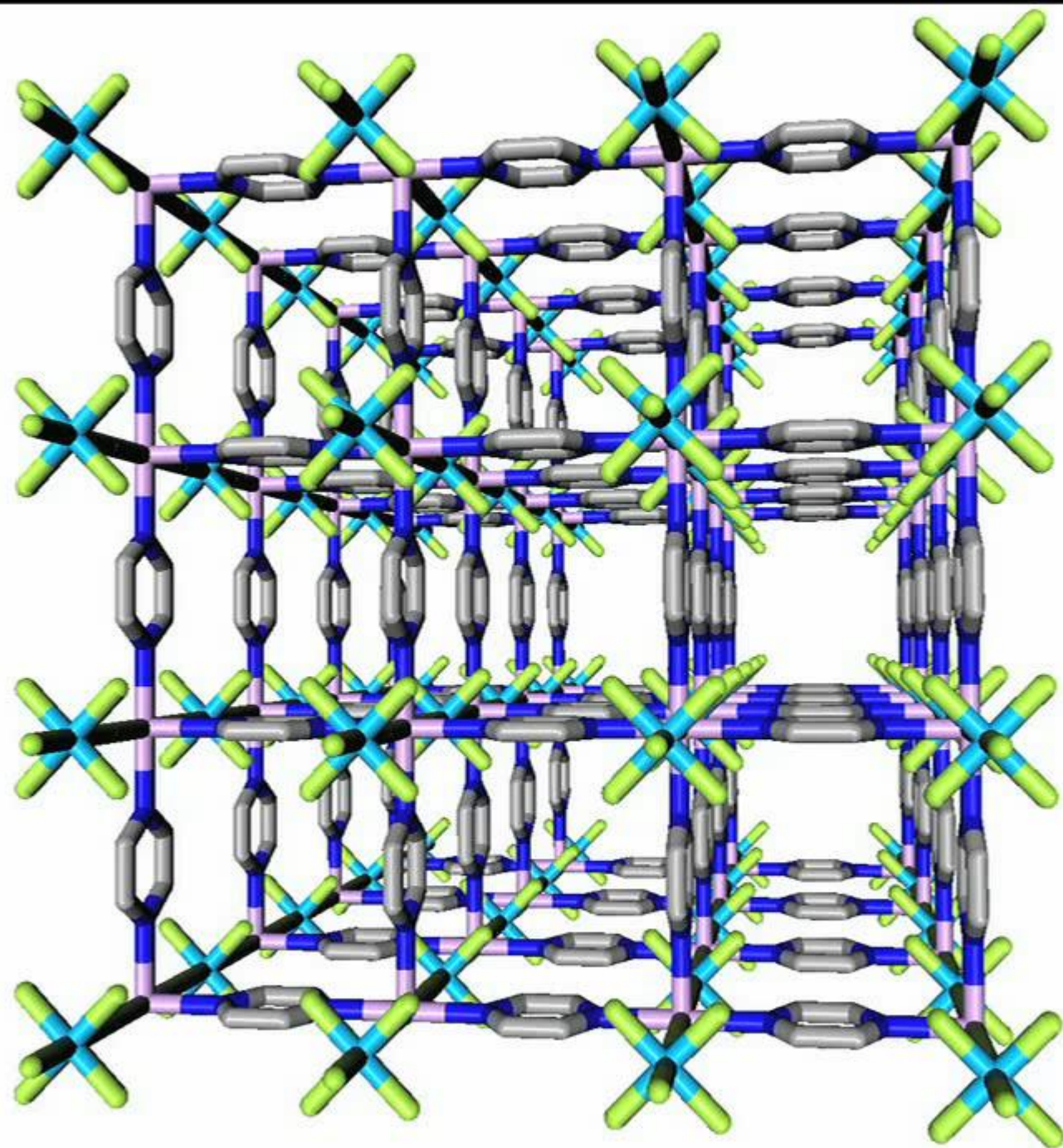
MOF  
MIL-101

Mixed Matrix membranes

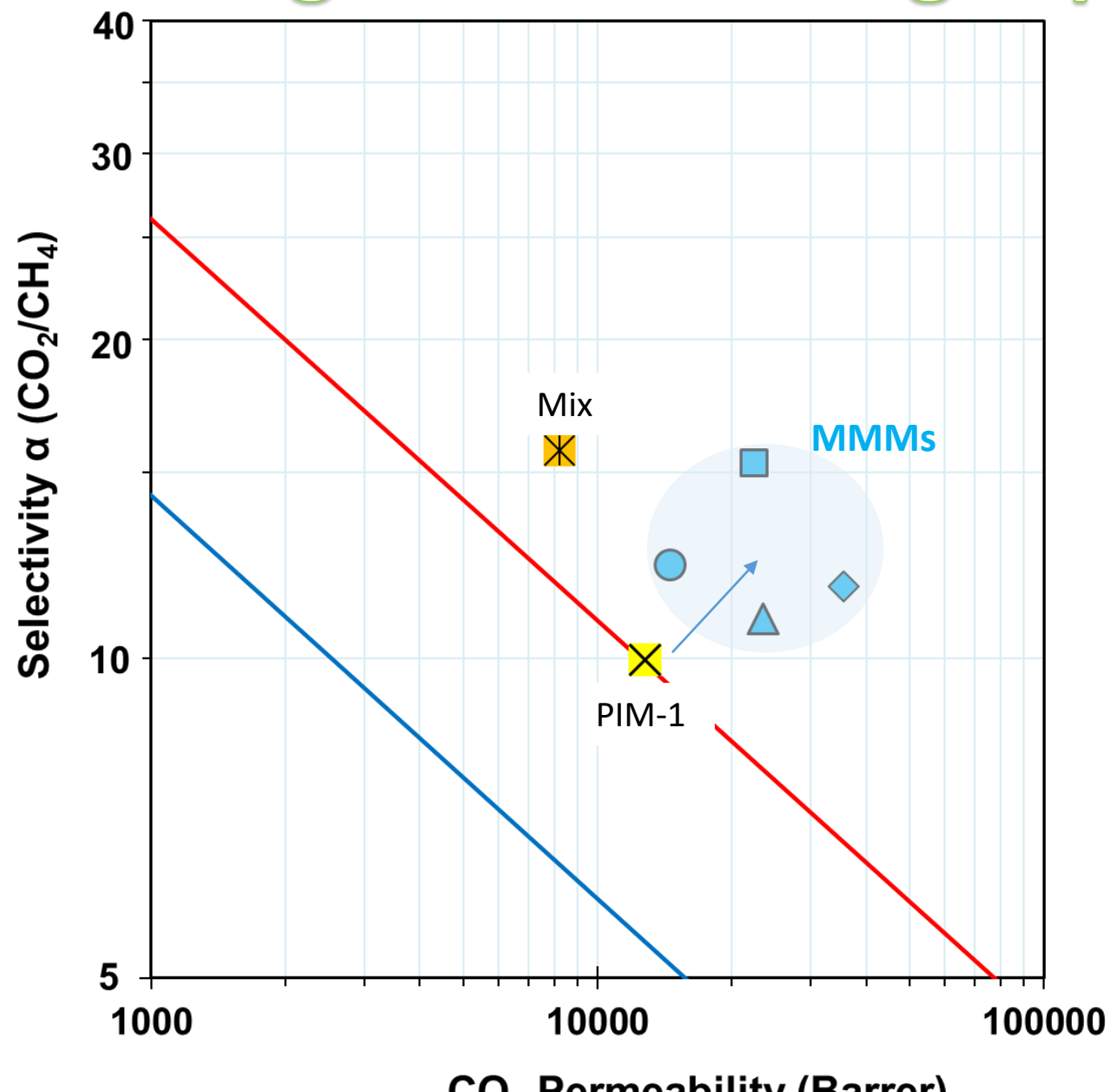
# Single gas permeation test



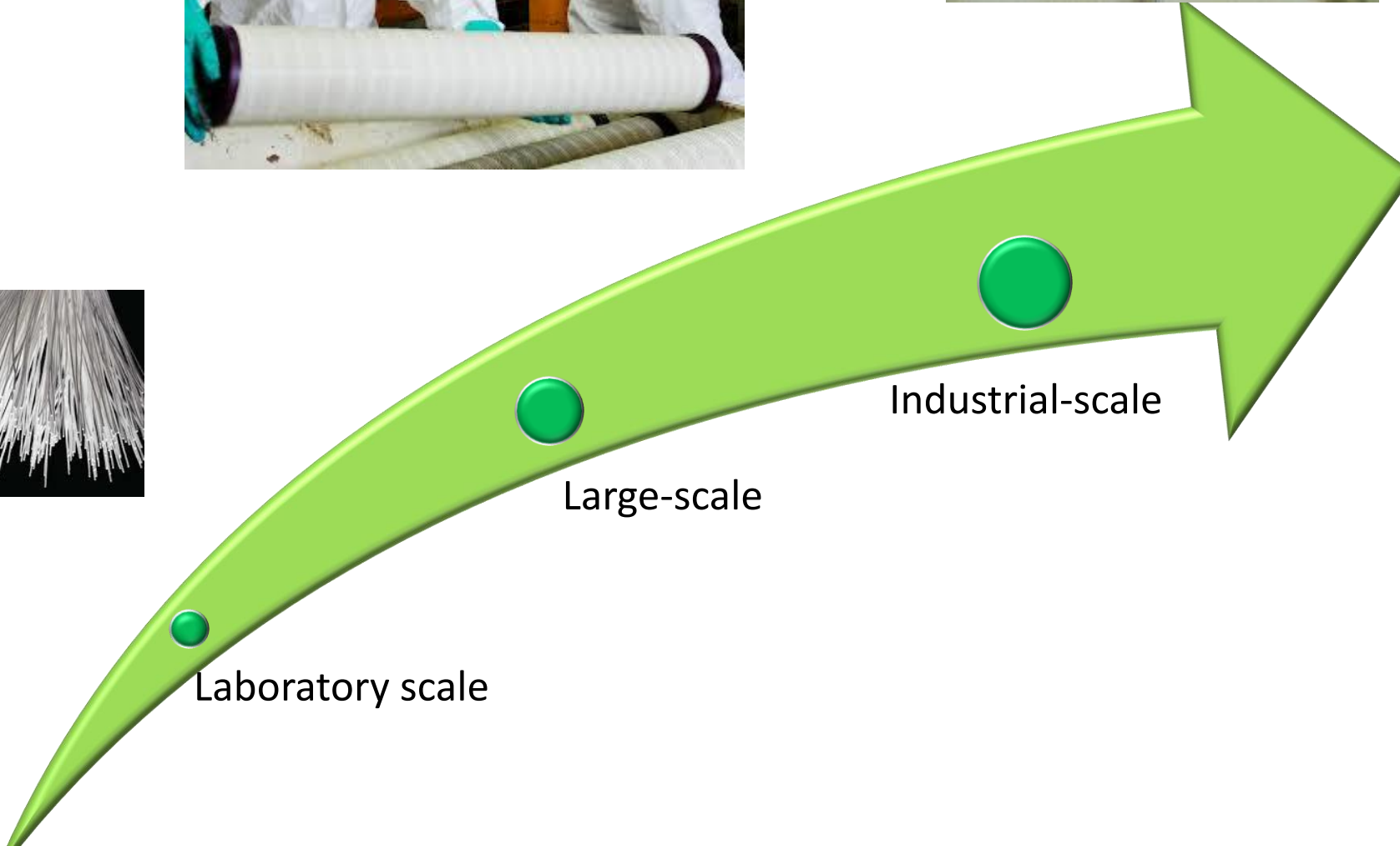




# Overview of Single and Mixed gas performance



# Scale-up



Laboratory scale



Large-scale



Industrial-scale



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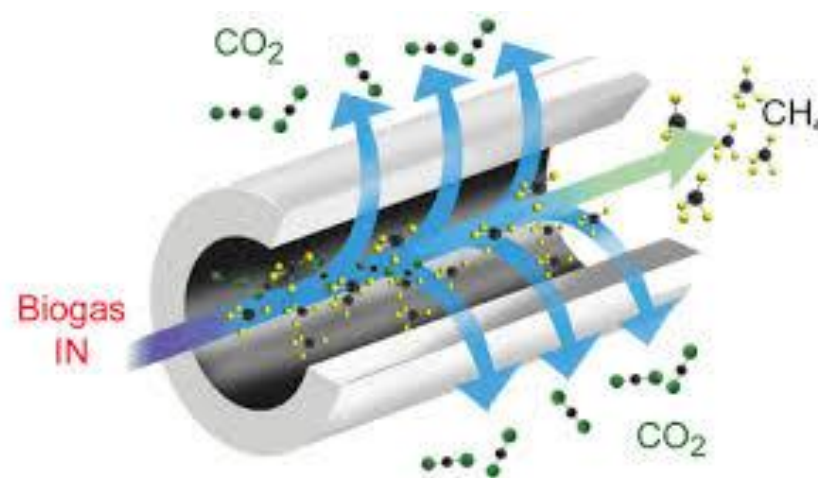


## Simultaneous production of biomethane and food grade CO<sub>2</sub> from biogas: an industrial case study†

Elisa Esposito,<sup>a</sup> Loredana Dellamuzia,<sup>b</sup> Ugo Moretti,<sup>b</sup> Alessio Fuoco,<sup>a</sup>  
Lidietta Giorno<sup>a</sup> and Johannes C. Jansen<sup>a</sup>

Cite this: *Energy Environ. Sci.*,  
2019, **12**, 281

Impact factor: 33

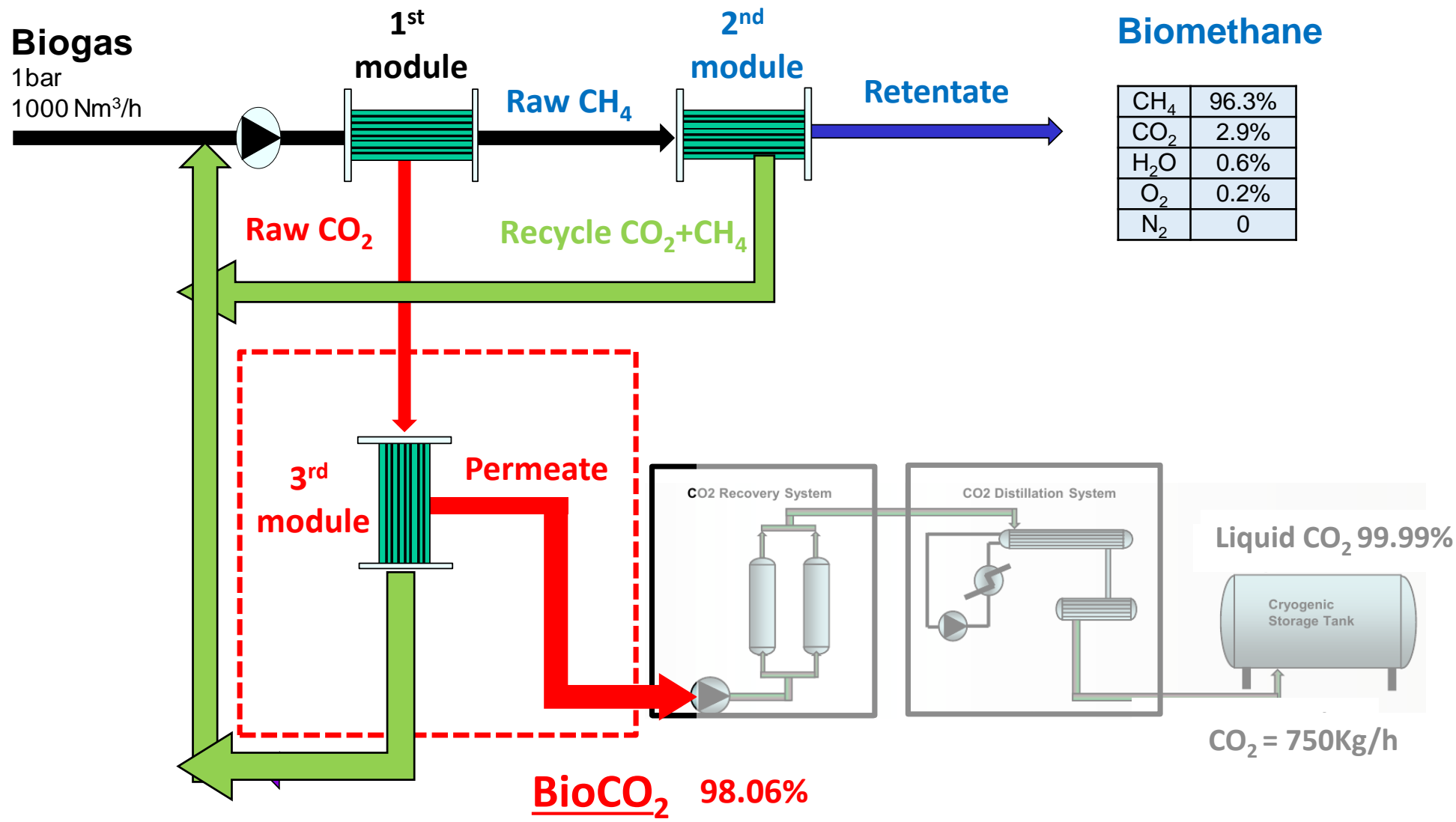


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PROJECT  
INDUSTRIALE**  
SIAD Group





# Three-stage membrane separation



Recycle 2<sup>nd</sup> permeate and 3<sup>rd</sup> retentate → minimum CH<sub>4</sub> loss



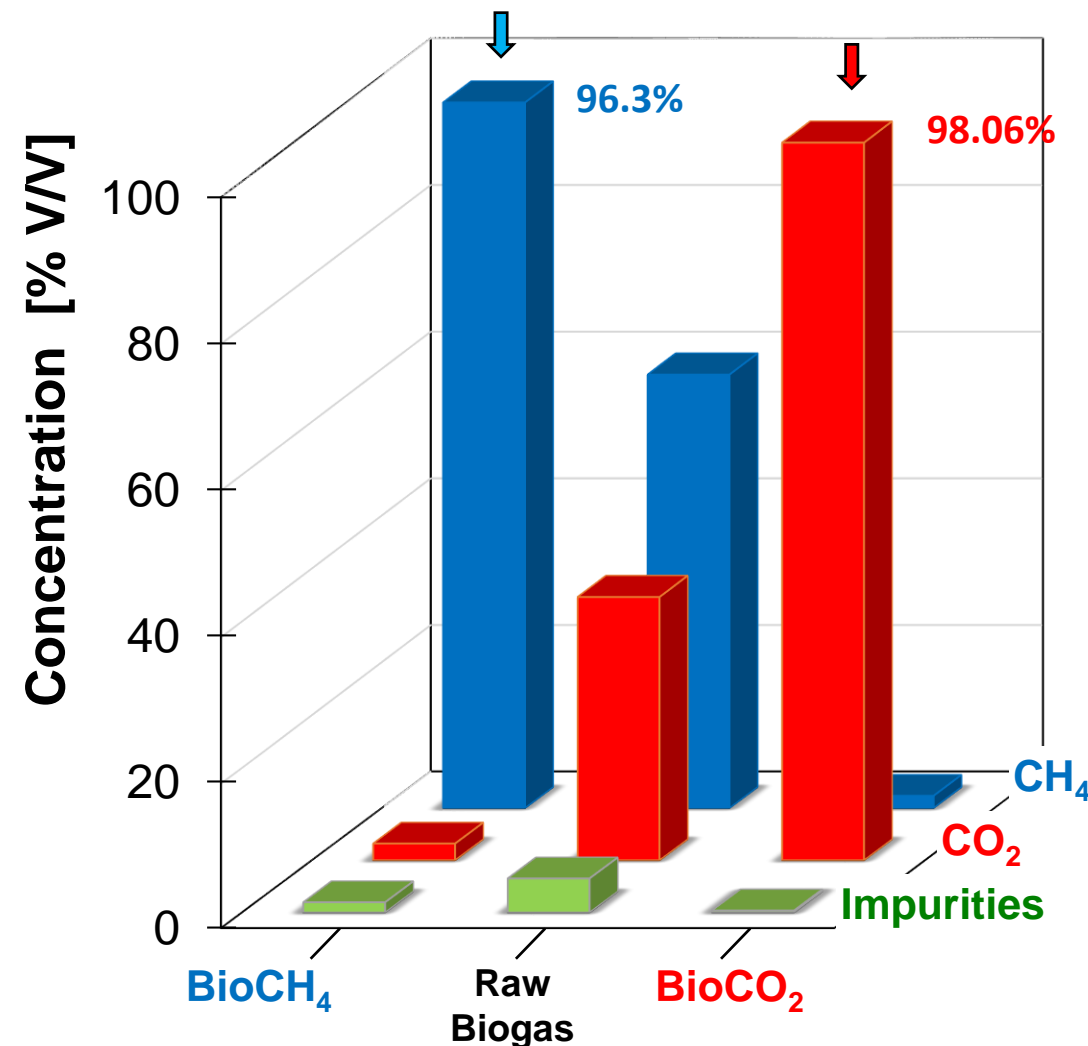
# Composition of Biomethane and BioCO<sub>2</sub> after membrane separation

→ Biomethane suitable for injection into the natural gas grid

Parameter	Limits allowed	Biomethane Membrane stage
Purity	>80%	96.3%
Wobbe Index (MJ/Sm <sup>3</sup> )	47.31-52.33	50.02
Density (g/cm <sup>3</sup> )	0.5548-0.8	0.56
Heating value (MJ/Sm <sup>3</sup> )	34.95-45.28	37.48

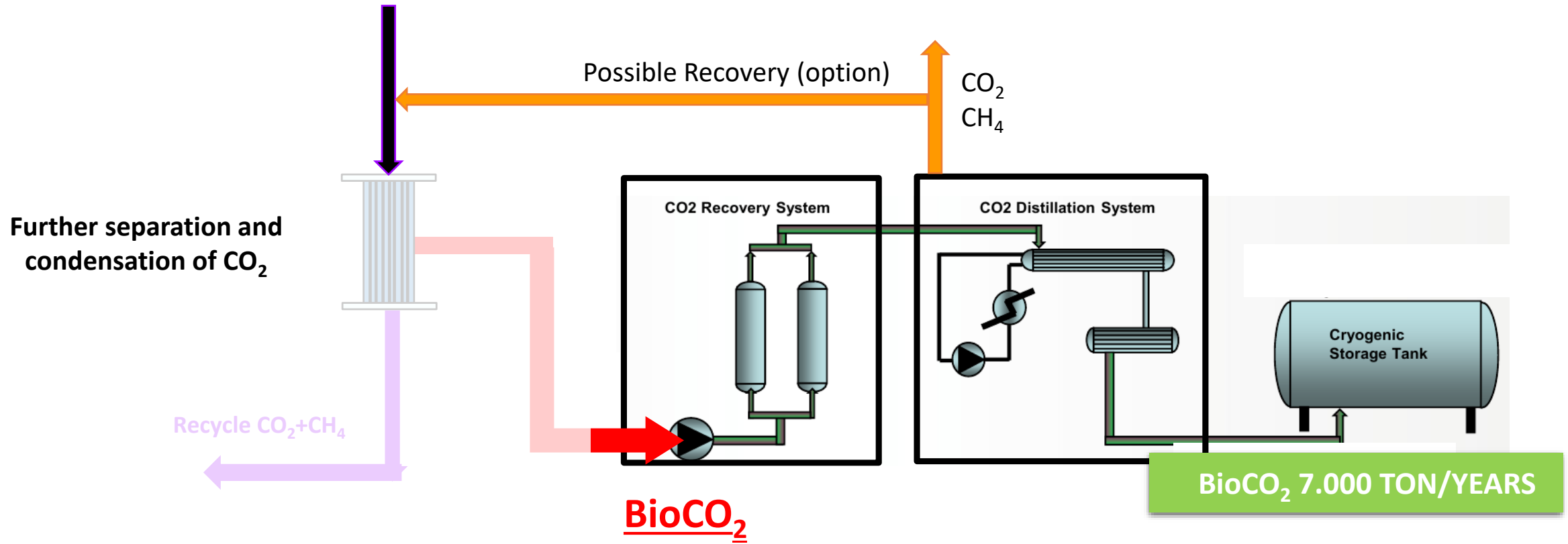
→ Further removal of trace impurities from CO<sub>2</sub> needed

Parameter	Limits EIGA/ISBT	CO <sub>2</sub> Membrane stage	Liquefied CO <sub>2</sub>
	Threshold	Measured values	Measured values
Purity	99.9% v/v max	98.06 v/v max	99.998% v/v
humidity	20 ppm v/v max	120 ppm v/v	-
acidity	comply with the test	comply with the test	comply with the test
Oxygen	30 ppm v/v max	200 ppm v/v	1.9 ppm v/v



# CO<sub>2</sub> recovery unit

Final CO<sub>2</sub> liquefying step



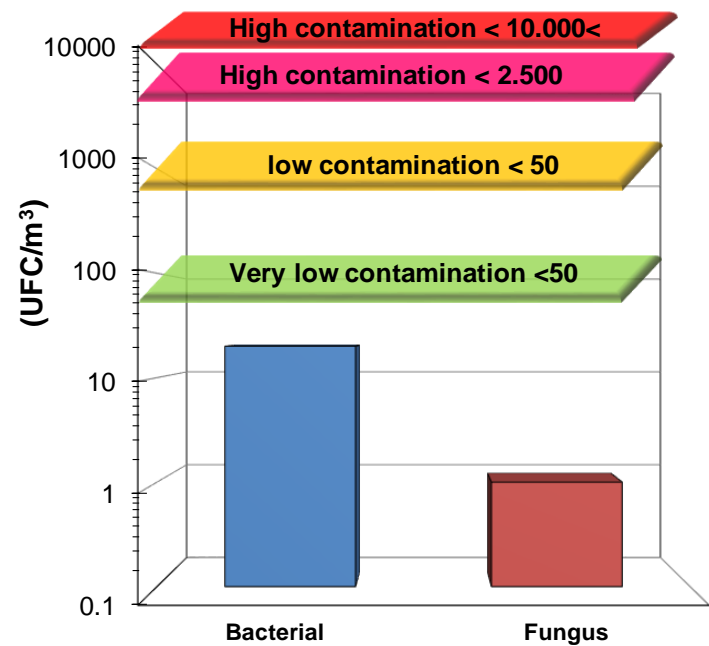
Additional advantage:

CH<sub>4</sub> recycle from this step back to the membrane separation unit leads to 100% CH<sub>4</sub> recovery and zero CH<sub>4</sub> emission

# Food grade BioCO<sub>2</sub>

The final liquefied CO<sub>2</sub> is **chemically and microbiologically pure**, respecting the limits of **food grade quality** proposed by the EIGA/ISBT

## BIOLOGICAL ANALYSIS



## CHEMICAL ANALYSIS

Parameter	Limits EIGA/ISBT	CO <sub>2</sub> Membrane stage	Liquefied CO <sub>2</sub>
	Threshold	Measured values	Measured values
Purity	99.9% v/v max	98.06 v/v max	99.998% v/v
humidity	20 ppm v/v max	120 ppm v/v	-
acidity	comply with the test	comply with the test	comply with the test
Oxygen	30 ppm v/v max	200 ppm v/v	1.9 ppm v/v





# Conclusions

Membrane processes produces simultaneously renewable energy and re-cycle of CO<sub>2</sub>

The energy cost of CO<sub>2</sub>/CH<sub>4</sub> membrane separation (0.3 kWh/m<sup>3</sup>) is lower compared to that the traditional separation techniques (0.6 kWh/m<sup>3</sup>).

3000 m<sup>3</sup>/h of Biomethane from organic waste that can be fed directly into the natural gas grid.

32.000 tonnes/year of CO<sub>2</sub> from a useless by-product to a food-grade quality gas for food and beverage industry

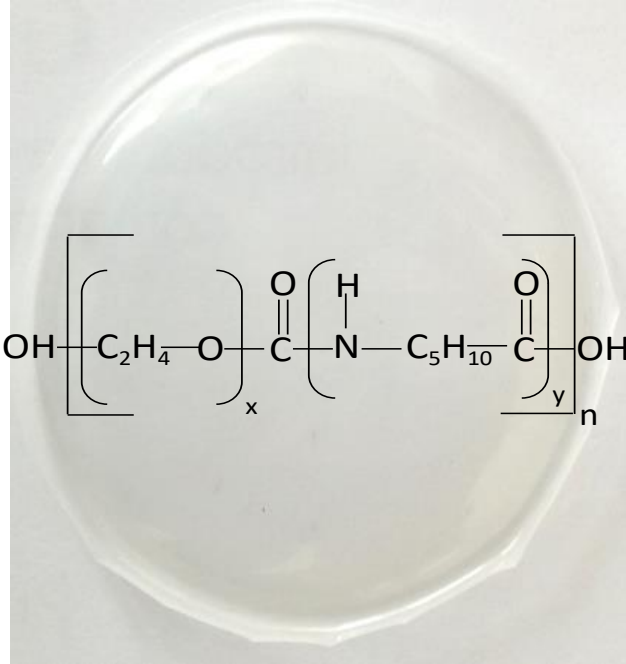
No CO<sub>2</sub> and CH<sub>4</sub> are released into the atmosphere and organic waste is consumed: first “carbon negative” plant in Italy

# Future challenges

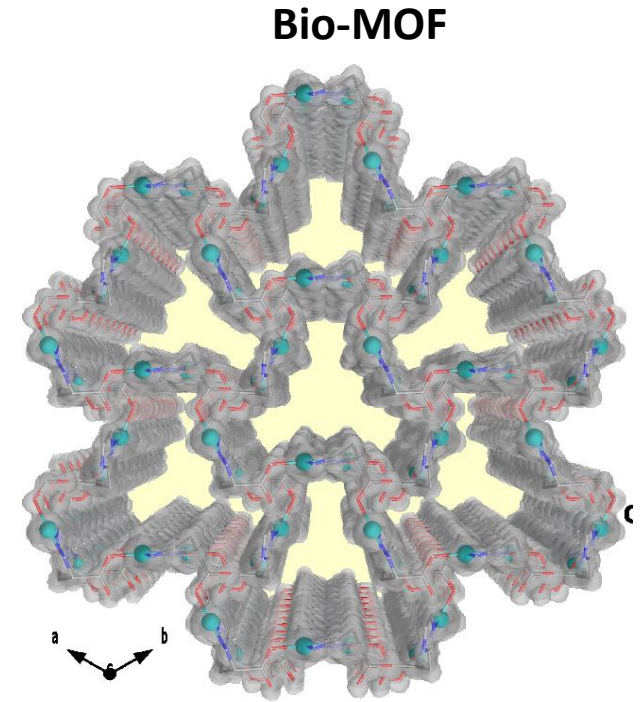


# Future challenges:

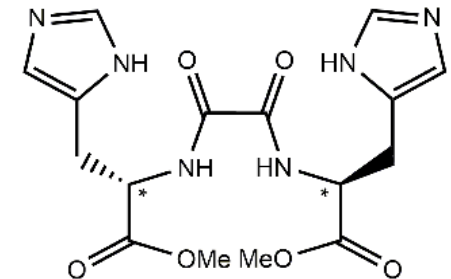
To green materials



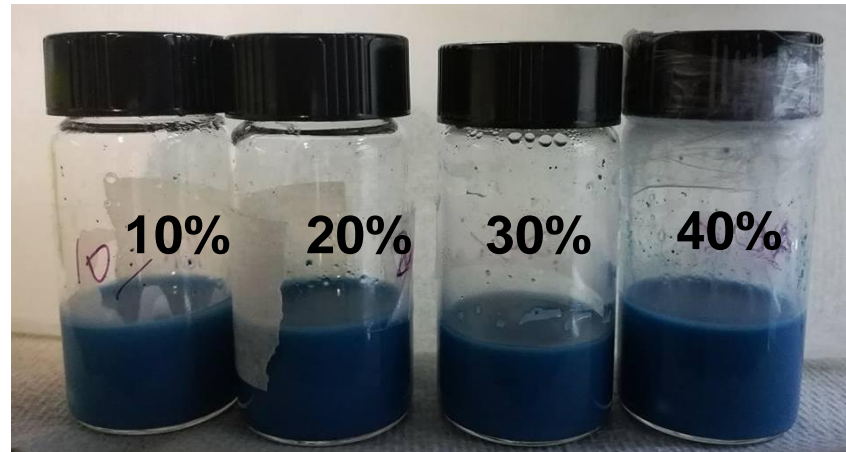
Pebax® membrane



Hismox-MOF



Histidin amino acid linker

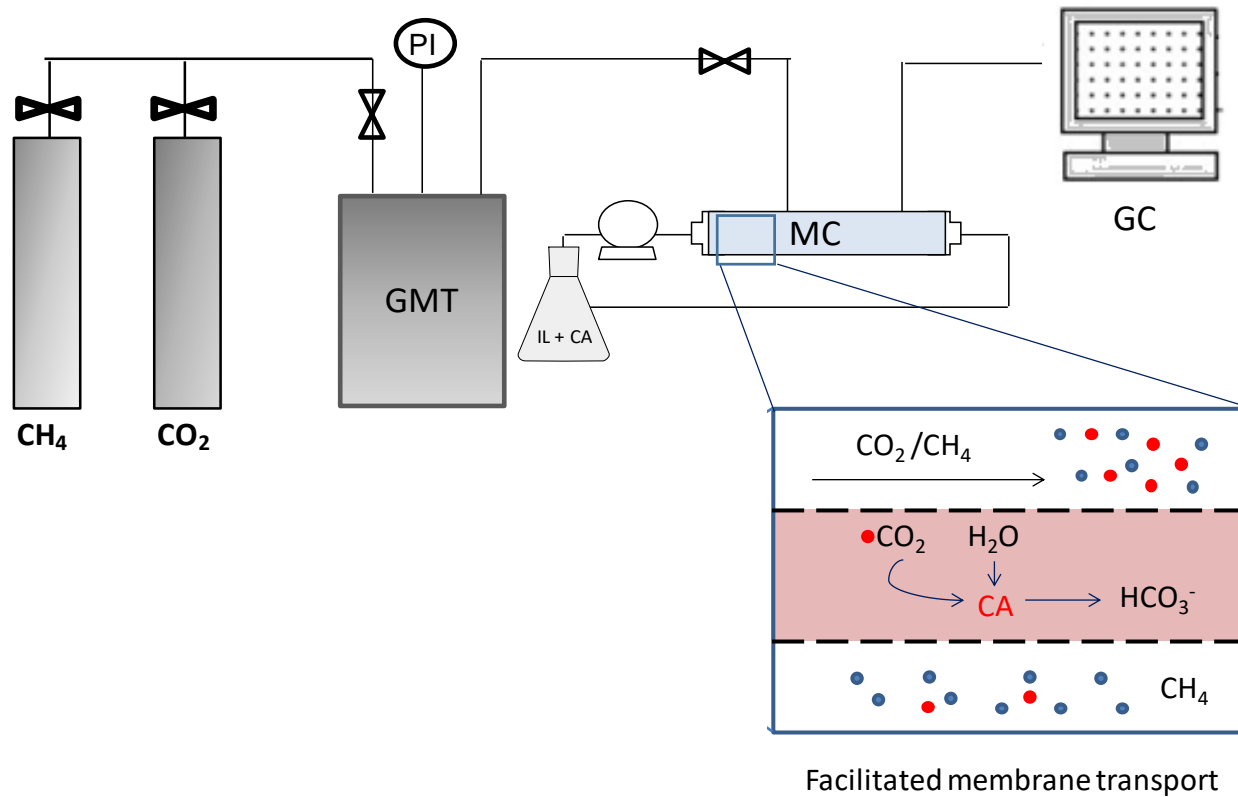


Green solvent: mixture of EtOH/H<sub>2</sub>O 70/30 vol%

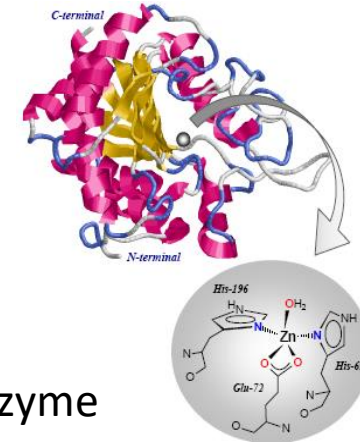
# Future challenges

## To Bio-process

Potential use of membrane contactors for CO<sub>2</sub>/CH<sub>4</sub> separation by facilitated CO<sub>2</sub> transport in Holo fiber membranes



The basic idea is to mimic the reaction that takes place inside the erythrocytes

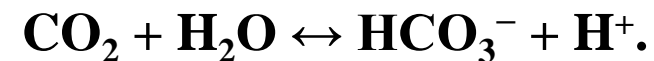


Enzyme

### Carboxylate anhydrase

- Metal protein
- Zinc functional group
- Catalyzes hydration of carbon dioxide

### Hydration of carbon dioxide



## RESEARCH ABROAD



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Thank you for your attention